

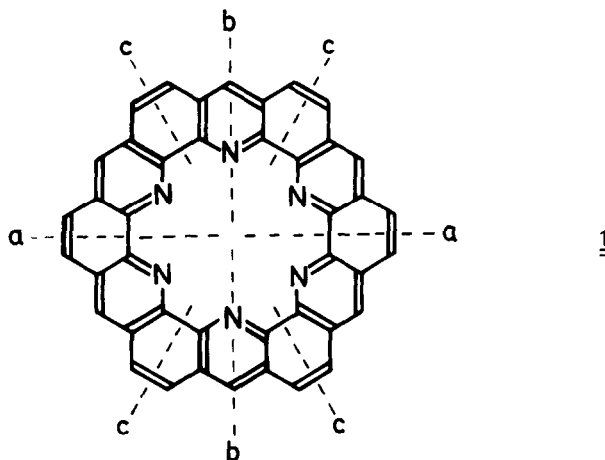
## EN ROUTE TO HEXAAZA-KEKULENE

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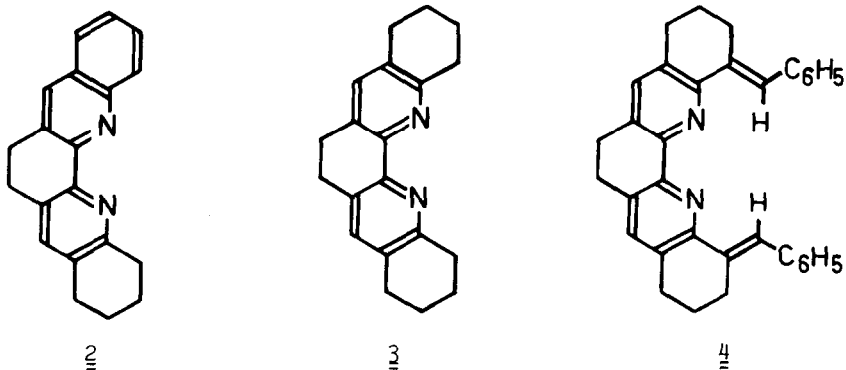
**Abstract:** As synthetic precursors of hexaaza-kekulene 1 the dibenzo[b,j]=[1,10]phenanthroline systems 2, 3, 4, 5 and 6 were prepared. From 6 via 7 a product was obtained for which the dodecahydrohexaaza-kekulene structure 8 is suggested on the basis of <sup>1</sup>H-NMR comparison with the dipyrido[3,2-c;=2',3'-h]acridine (14) for which two syntheses are reported.

The first member of the cycloarene family kekulene (cyclo[d.e.d.e.d.e.=d.e.d.e.d.e]dodecakisbenzene) has previously been synthesized in our group <sup>1)</sup>, and its molecular structure and spectroscopic properties were determined <sup>2)</sup>. Of the heteroatomic analogues of kekulene so far only a dodecahydro-18,21-dioxonia-kekulene has been synthesized <sup>3)</sup>. 19,20,21,22,23,24-Hexaaza-kekulene (1) with six coplanar nitrogen atoms forming the central cavity is of special interest (e.g., regarding its complexation behaviour). The synthesis of 1 has been tried following the strategies outlined as a, b, and c in formula 1. In this paper, we report on results achieved along route b.

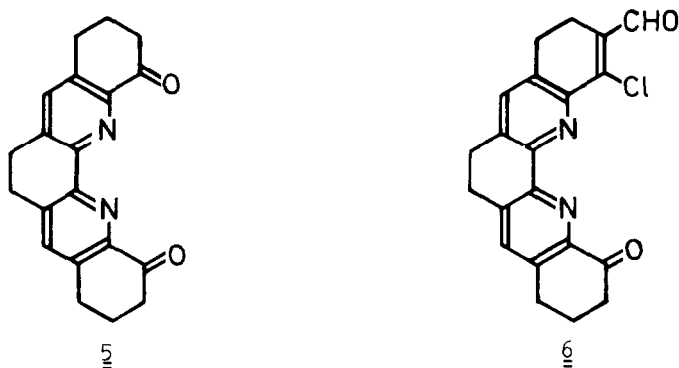


2,3,4,5,7,8-Hexahydrodibenzo[b,j][1,10]phenanthroline (2) was prepared from 1,2,3,4,5,6,7,8-octahydroacridin-4-one following the two-step synthesis of Tilichenko et al. <sup>4)</sup> with optimizing variations. The latter <sup>4,5)</sup> was prepared from 1,2,3,4,5,6,7,8-octahydroacridine in two steps in analogy to a

known procedure <sup>7)</sup> for the synthesis of 9. From 2 by catalytic hydrogenation (PtO<sub>2</sub>, trifluoroacetic acid, 3.3 at H<sub>2</sub>, 32 h, 20°C), 2,3,4,5,7,8,10,11,12,13-decahydrodibenzo[b,j][1,10]phenanthroline (3) <sup>5)</sup> was obtained (~ quant., m.p. 163° -164°C). Condensation of 3 with benzaldehyde (acetic anhydride, 9.5 h, reflux) yielded the 2,13-dibenzylidene derivative 4 <sup>5)</sup> (m.p. 208°C; 61%) which was ozonized in dichloromethane/methanol at -78°C to the diketone 5 <sup>5)</sup> (m.p. 255°C, dec. 90%).

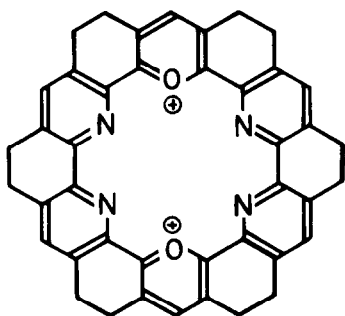
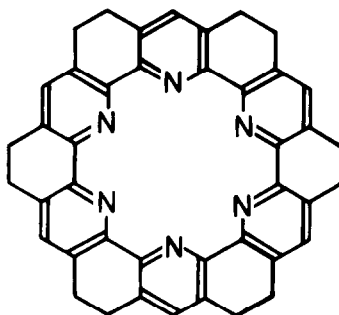


For the Vilsmeier reaction of 5 to 6, a solution of 5 in dry dichloromethane was added at 0 - 5°C to a reaction mixture prepared from dimethylformamide in dry dichloromethane by dropwise addition of freshly distilled phosphorous oxychloride. After 48 h stirring at 20°C and aqueous work up the  $\beta$ -chlorovinylaldehyde 6 <sup>5)</sup> (m.p. 330°C, dec.) was obtained in 84% yield.

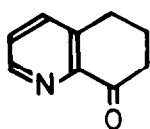
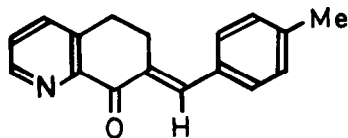
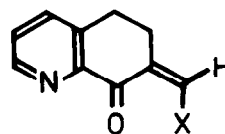


Cyclisation of 6 to the dodecahydro-19,20,22,23-tetraaza-21,24-dioxonia-kekulene 7 was achieved with perchloric acid/acetic acid (2.5 h, 20°C). The extremely insoluble product isolated (dec. > 300°C; ~ 94%), according to elemental analysis, is 7-tetra-perchlorate. Under corresponding conditions 7-tetraakis(tetrafluoroborate) <sup>5)</sup> was obtained. Reaction of 7-tetra-perchlorate with ammonia in acetonitrile (18 h, reflux) yielded, by dichloromethane extraction of the residue obtained after evaporation of the acetonitrile, a product as pale-yellow powder in 3% yield. Mass spectra [CI: m/z 617 (20%, [M - H]<sup>+</sup>),

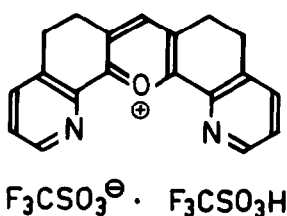
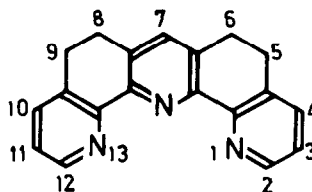
FAB:  $m/z$  619 (70%,  $[M + H]^+$ ) do not rule out structure 7. Of more significance, however, is the  $^1\text{H-NMR}$  spectrum (360 MHz,  $\text{CD}_2\text{Cl}_2$ ) which shows only two singlets at 3.090 (4H) and 7.599 (1H) as expected for the  $D_{6h}$ -symmetrical dodecahydro-19,20,21,22,23,24-hexaaza-kekulene 8.

78

Since, so far, the assumption of structure 8 is essentially based on  $^1\text{H-NMR}$ , the comparison of the  $\delta$ -values obtained for 8 with those of 5,6,8,9-tetrahydrodipyrido[3,2-c;2',3'-h]acridine (14) is of special interest. 14 was synthesized from 5,6,7,8-tetrahydroquinolin-8-one (9) in two different ways: By condensation of 9 with p-tolualdehyde (piperidine, ethanol; 2 h, reflux) 7-(4-methylbenzylidene)-5,6,7,8-tetrahydroquinolin-8-one (10)<sup>5)</sup> was obtained (m.p. 176 - 177°C; 67%) which with 9 on heating with trifluoromethanesulfonic acid (5 h, 100°C) yielded 5,6,8,9-tetrahydrodipyrido[3,2-c;2',3'-h]xanthylum trifluoromethanesulfonate 13<sup>5)</sup> (m.p. 118°C; 29%). From 13 on heating with aqueous ammonia/ethanol (1 h, reflux) 14<sup>5)</sup> was obtained (m.p. 229°C; 21%). For an alternative synthesis of 14 from 9, 7-hydroxymethylene-5,6,7,8-tetrahydroquinolin-8-one (11) was prepared from 9 by condensation with ethyl formate (EtONa, ether; 12 h, 20°C; 94%); 11<sup>5)</sup> (m.p. 106°C) was converted by reaction with ammonia in chloroform into the aminomethylene derivative 12 which without purification was reacted with 9 (ammonium acetate, 12 h, 120°C) to yield 14 in 22% yield. Two independent syntheses of the hitherto unknown 14 were very recently reported<sup>6)</sup>.

910

11: X = OH  
12: X = NH<sub>2</sub>

1314

The  $^1\text{H-NMR}$  spectrum (360 MHz,  $\text{CD}_2\text{Cl}_2$ ) shows the following signals:  $\delta =$  3.009 ('s', 8H; 5,6,8,9-H), 7.230 (dd,  $J = 7.6$  and  $4.7$  Hz, 2H; 3,11-H), 7.451 (s, 1H; 7-H), 7.579 (dd,  $J = 7.6$  and  $1.5$  Hz, 2H; 4,10-H), 8.683 (dd,  $J = 4.7$  and  $1.5$  Hz, 2H; 2,12-H). The excellent agreement of the signals for 7-H and 5,6,8,9-H of 14 with the two signals of 8 are a strong indication for the structure suggested. - Dehydrogenation to hexaaza-kekulene 1 is in progress.

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- 1) H. A. Staab and F. Diederich, Chem. Ber. 116, 3487 (1983); preliminary commun.: F. Diederich and H. A. Staab, Angew. Chem. 90, 383 (1978); Angew. Chem. Int. Ed. Engl. 17, 372 (1978).
  - 2) H. A. Staab, F. Diederich, C. Krieger and D. Schweitzer, Chem. Ber. 116 3504 (1983); preliminary commun.: C. Krieger, F. Diederich, D. Schweitzer and H. A. Staab, Angew. Chem. 91, 733 (1979); Angew. Chem. Int. Ed. Engl. 18, 699 (1979); cf. D. Schweitzer, K. H. Hausser, H. Vogler, F. Diederich and H. A. Staab, Mol. Physics 46, 1141 (1982).
  - 3) A. R. Katritzky and C. M. Marson, J. Am. Chem. Soc. 105, 3279 (1983).
  - 4) G. A. Klimov, V. A. Stonik and M. N. Tilichenko, Khim. Geterotsikl. Soedin. 6, 821 (1973), CA 79: 92057f; see also G. A. Klimov and M. N. Tilichenko, Khim. Geterotsikl. Soedin. 1, 306 (1967), CA 70: 87539z.
  - 5) Correct elemental analyses and spectroscopic data are in accordance with the structures suggested.
  - 6) R. P. Thummel and Y. Jahng, J. Org. Chem. 50, 2407 (1985).
  - 7) E. Reimann and H.-L. Ziegler, Liebigs Ann. Chem. 1976, 1351.  
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